

## A FACILE SYNTHESIS OF ANTITUMOR AGENT BATRACYLIN VIA INTRAMOLECULAR AZA-WITTIG REACTION

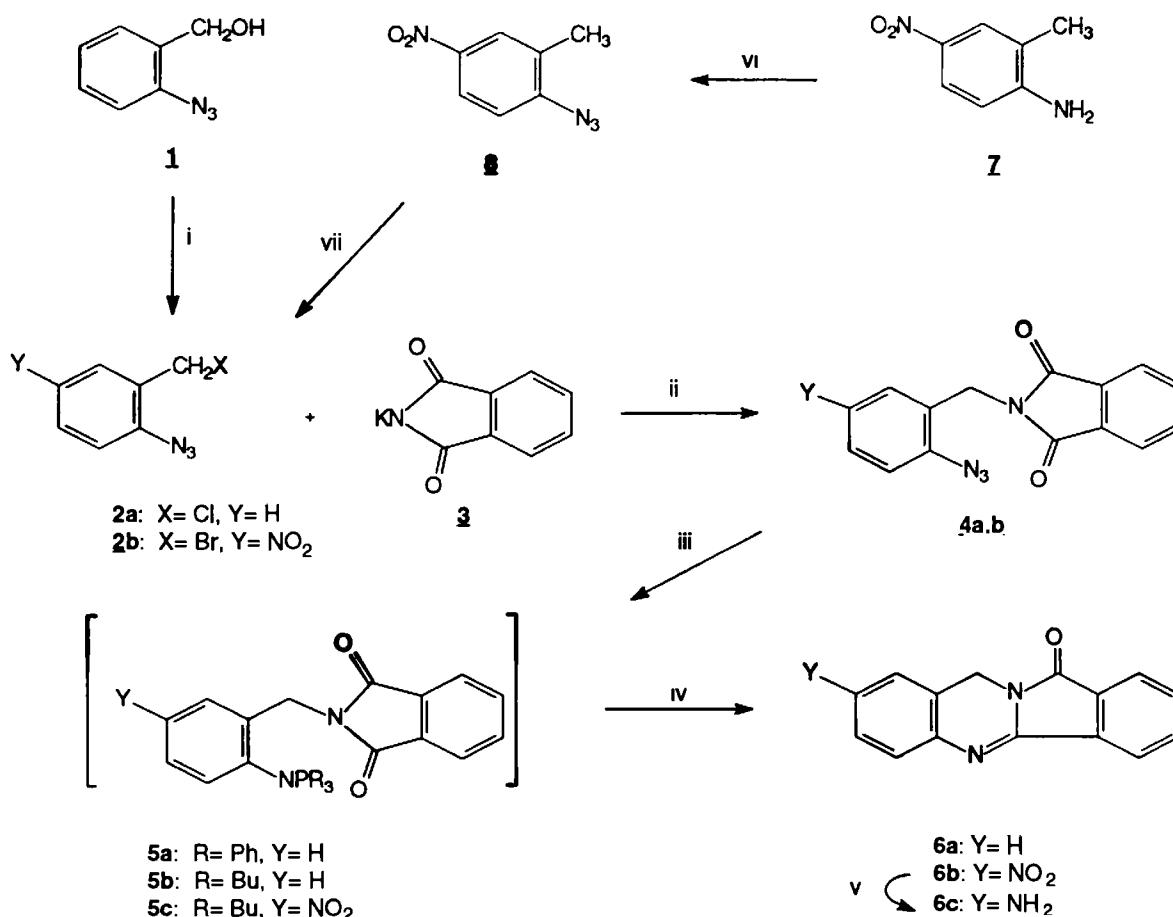
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**Abstract :** A facile synthesis of antitumor agent Batracyclin (NSC-320846), 8-aminoisoindolo[1,2-b]-quinazolin-12(10H)-one by intramolecular aza-Wittig methodology is reported.

The aza-Wittig methodology in heterocyclic synthesis has been growing quite rapidly in recent years, however, application to synthesis of various types of ring systems seems to be yet a challenging problem, because the cyclization reactivity in the intramolecular version is governed by many factors such as ring-size, substituents on P and N, and carbonyl groups (1). We have reported recently novel and efficient route for synthesis of oxazoles (2), imidazolinones (3a), iminolactams (4), quinazolinones (3,5), and 1,4-benzodiazepinones (6) via intramolecular aza-Wittig reactions. In this communication we wish to report a facile synthesis of dihydroquinazoline ring system by a consecutive Staudinger/aza-Wittig reaction of N-(*o*-azidobenzyl)phthalimides using triphenylphosphine or tributylphosphine, and application to a facile synthesis of antitumor agent Batracyclin (NSC-320846) **6c** (7).

The reaction of *o*-azidobenzyl chloride **2a** prepared from the known azidoalcohol **1** (8) with potassium phthalimide **3** in DMF at 55 °C for 1 h gave the azido precursor **4a** in 96% yield as a faintly yellowish crystal after usual work up and flash chromatography on a silica gel column (hexane/AcOEt, 4/1) (Scheme). The reaction of **4a** with triphenylphosphine (1.1 equiv.) in dry xylene at room temperature for 5 h proceeded cleanly to generate iminophosphorane **5a** via the Staudinger reaction but the cyclization via the intramolecular aza-Wittig reaction was very sluggish as judged from TLC. By heating to 90 °C for 9 h, however, the cyclization was completed to afford 10,11-dihydroiso-indolo[1,2-b]quinazolin-12-one **6a** quantitatively after flash chromatography (silica gel, hexane/AcOEt, 2/1) as a yellowish crystal. The same reaction of **4a** with tributylphosphine (1.1 equiv.) in dry benzene proceeded more rapidly and after



## Scheme

12 h at room temperature (ca.  $20-25^\circ\text{C}$ ), **6a** was obtained in 97% isolated yield. Above results confirmed that the fused dihydroquinazoline ring could be synthesized efficiently in one-pot via the consecutive Staudinger/aza-Wittig reaction.

We applied this methodology to an improved synthesis of the well known synthetic antitumor agent Batracylin (NSC-320846) **6c** (**7**). Commercially available 2-methyl-4-nitroaniline **7** was converted to

azide **8** (87%) by the standard procedure, and benzylic bromination of **8** with NBS in dry benzene gave azidobenzyl bromide **2b** in 72% yield as a brownish solid. The reaction of **8** with potassium phthalimide in DMF at 80 °C for 6 h gave **4b** in 70% yield as a yellowish crystal after flash chromatography (silica gel, hexane/AcOEt, 2/1). Treatment of **4b** with tributylphosphine (1.1 equiv.) in dry xylene for 12 h at 140 °C afforded **6b** (86%) as a yellowish crystal after usual work up. Such heating was required for this cyclization because the electron withdrawing nitro substituent at para-position to the imino group diminished reactivity of the imino phosphorane. Reduction of the nitro group with ammonium formate and 10% Pd-C in CH<sub>2</sub>Cl<sub>2</sub>-MeOH afforded amino derivative, Batracyclin **6c** in 98% yield as a yellowish solid which was characterized by spectral and analytical data (Scheme).

The known synthetic routes to Batracyclin use a) thermal condensation of 2,5-diaminobenzylamine with phthalic anhydride (56%) (**9**), and b) acid catalyzed cyclization of N-[2-ethoxycarbonylamino-5-acetylaminobenzyl]phthalimide in conc. H<sub>2</sub>SO<sub>4</sub> at 100 °C (95%) (**10**). Compared with these two known methods, present synthesis provides a new convenient route to Batracyclin and fused dihydroquinazoline derivatives from readily available precursors.

Satisfactory spectroscopic and analytical data have been obtained for all compounds:

**4a**: mp 184–186°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) δ 7.92–7.70 (m, 4H), 7.36–7.03 (m, 4H), 4.86 (s, 2H); MS(EI) m/z(%) 278(M+, 19), 250(50), 249(64), 104(100).

**6a**: mp 186.5–188.0°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) δ 8.10–7.17 (m, 8H), 5.01 (s, 2H); MS(EI) m/z(%) 234 (M+, 65), 233(100).

**2b**: mp 55–58°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) δ 8.30 (dd, 1H, J=0.4, 2.6 Hz), 8.24 (dd, 1H, J=2.6, 8.6 Hz), 7.30 (d, 1H, J=8.6 Hz), 4.48 (s, 2H); MS(EI) m/z(%) 256 (M+, 13), 230(21), 149(100).

**4b**: mp 184–188°C (dec.); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) δ 8.22 (dd, 1H, J=2.6, 8.8 Hz), 8.08 (d, 1H, J=2.6 Hz), 7.29 (d, 1H, J=8.8 Hz), 7.94–7.76 (m, 4H), 4.89 (s, 2H); MS(EI) m/z(%) 323 (M+, 10), 295(40), 104(100).

**6b**: mp 283–287 (AcOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 8.21 (dd, 1H, J=2.7, 8.5 Hz), 8.10–8.08 (m, 2H), 7.96–7.94 (m, 1H), 7.80–7.73 (m, 2H), 7.61 (d, 1H, J=8.5 Hz), 5.08 (s, 2H); MS(EI) m/z(%) 279 (M+, 100).

**6c**: mp 244–247°C, lit. 287°C, lit. 287–288°C (16), 270°C (dec.) (17); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 8.02 (d, 1H, J=7.5 Hz), 7.90 (d, 1H, J=7.5 Hz), 7.69 (ddd, 1H, J=7.5, 7.5, 1.0 Hz), 7.63 (ddd, 1H, J=7.5, 7.5, 1.0 Hz), 7.32 (d, 1H, J=8.5 Hz), 6.63 (dd, 1H, J=8.5, 2.5 Hz), 6.50 (d, 1H, J=2.5 Hz), 4.92 (s, 2H), 3.86 (s, 2H, NH<sub>2</sub>); MS(EI) m/z(%) 249 (M+, 85), 248 (100).

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